

Creating Sustainable Concrete Through Carbon Sequestration

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Introduction

Background

The construction industry is one of the largest sources of carbon dioxide (CO₂), with concrete production accounting for 8% of global emissions (Rodgers, 2018). In fact, concrete emits 0.8-0.95 tonnes of CO₂ for every tonne produced, totaling over 1.5 billion tonnes of CO₂ annually (Massoumi Nejad et al., 2025). The process of carbonation immobilizes CO₂ as stable carbonates via dissolution and recombination with calcium or magnesium (Li et al., 2022). This presents a promising solution for CO₂ sequestration in concrete (Laurent, 2023).

Concrete Carbonation

Passive Carbonation: Portland cement (concrete binder) naturally carbonates from atmospheric CO₂ due to moisture content and calcium-bearing phases (portlandite (Ca(OH)₂) and Calcium Silicate Hydrate) (Anabela et al., 2020)

Carbonation Curing: Exposing fresh concrete to controlled CO₂ in a pressurized carbonation chamber accelerates the carbon sequestration process (Li et al., 2022)

Mineral Carbonation: Pretreat Ca/Mg rich Supplementary Cementitious Materials (SCM's) with CO₂ and then use in concrete (Pu et al., 2021 and Kazemian & Shafei, 2023)

Passive Carbonation

Figure 1: Passive carbonation diagram (Possan et al., 2017)

- Stable, long-term CO₂ storage (Anabela et al., 2020)
- Long exposure periods (Van Roijen et al., 2024)

Carbonation Curing

Figure 2: Carbonation curing setup (Shao et al., 2014)

- Rapid CO₂ uptake
- Increased strength
- Process-dependent and inconsistent (Li et al., 2022)

Mineral Carbonation

Figure 3: Mineral carbonation flowchart (Ragipani et al., 2022)

- Stable carbonates
- Sustainable SCM's
- Material-dependent
- Strength tradeoffs (Kazemian & Shafei, 2023)

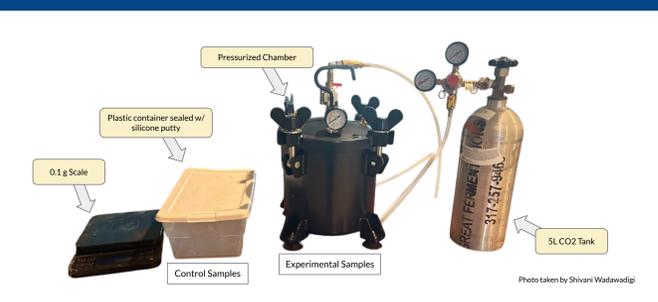
Problem

Existing CO₂ sequestration methods each provide benefits, but none solve all major challenges of speed, process consistency, and consistent material performance.

Engineering Goal

Create and optimize a structurally sound concrete mix from recycled materials that quickly and consistently sequesters CO₂ when carbonation-cured, helping mitigate carbon emissions and assist in the future of sustainable construction.

Experimental Setup



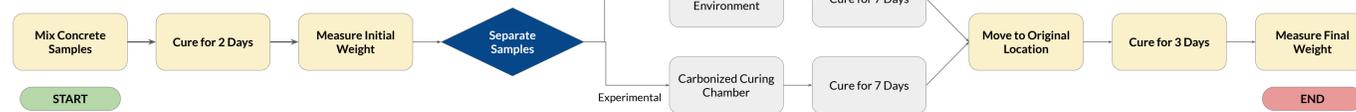
Materials & Methodology

Carbonation Testing

Three trials; Trials 1 and 2 tested the original 8 mixes while Trial 3 tested the optimized 9th mix.

- Mixes designed using an L9 Taguchi Array (4 Factors, 3 Levels) (see Table 2)
- CO₂ sequestration was measured via weight gain
- 8 mixes (see Table 1) tested in CO₂ pressure chamber
- RH controlled using a supersaturated NaCl solution to target an RH of 74-76% (Saturated Salt Solutions and Air Humidity, 2014)
- Temperature was maintained at ambient conditions
- Samples sealed to prevent dilution/contamination

Carbonation Experiment Workflow



Structural Testing

Measured through compression testing

- 32 samples; 4 samples per mix (2 control, 2 experimental)
- Tested MTS Insight Electromechanical Testing System until failure
- Peak load recorded

Optimization

Goal: maximize CO₂ Sequestration
Used a regression using JMP Student Version, shown in Figure 5.

Optimal Mix:

- Hydrated lime
- Steel Slag
- Water/Binder = 0.3

Mix	SCM	Aggregate	Additive	Water-Binder
Mix 1	Fly Ash	Recycled	None	0.4
Mix 2	Slag	Natural	Biochar	0.4
Mix 3	Hydrated Lime	Steel Slag	Accelerator	0.35
Mix 4	Fly ash	Steel Slag	Biochar	0.4
Mix 5	Slag	Recycled	None	0.4
Mix 6	Hydrated Lime	Natural	Accelerator	0.35
Mix 7	Fly Ash	Natural	Biochar	0.35
Mix 8	Slag	Steel Slag	Accelerator	0.45

Table 1. Initial Eight Concrete Mixes

Factors	Level 1	Level 2	Level 3
SCM	Fly ash	Slag	Hydrated Lime
Aggregates	Recycled	Natural	Steel Slag
Additives	None	Biochar	Accelerator
Water-Binder	0.35	0.4	0.45

Table 2. L9 Taguchi Array showing factors and levels used

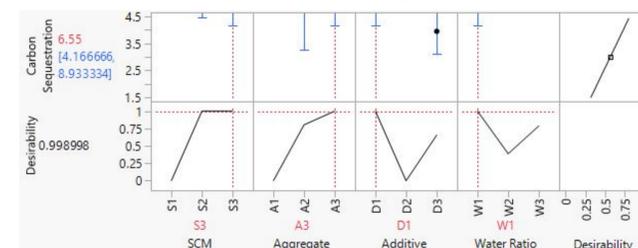


Figure 4. JMP Optimization of Concrete Mixes

Results and Discussion

Mix	Trial 1		Trial 2		Average	
	Avg. CO ₂ (%)	Avg. CO ₂ (g)	Avg. CO ₂ (%)	Avg. CO ₂ (g)	CO ₂ (%)	CO ₂ (g)
Mix 1	4.53%	2.90	5.89%	4.15	5.21%	3.52
Mix 2	3.39%	2.35	4.10%	2.95	3.74%	2.65
Mix 3	3.78%	3.55	4.46%	4.35	4.12%	3.95
Mix 4	2.19%	2.00	3.48%	3.25	2.84%	2.62
Mix 5	4.69%	3.15	5.12%	3.75	4.91%	3.45
Mix 6	2.38%	1.70	4.23%	2.80	3.30%	2.25
Mix 7	3.96%	2.75	4.82%	3.45	4.39%	3.10
Mix 8	3.57%	3.80	3.95%	3.95	3.76%	3.88

Table 3. Carbon Sequestration of Original Eight Mixes

Mix	Trial 3	
	Avg. CO ₂ (%)	Avg. CO ₂ (g)
Mix Optimal	6.60%	6.40

Table 4. Carbon Sequestration of Optimized Mix

Compression Test Results

- 32 compression tests (2 experimental, 2 control per mix)
- One experimental sample of mix 1 invalidated due to test failure (31 total results)
- 7 of the 8 mixes performed better than their control counterparts.
- 3 out of 8 mixes met industry standards of 4000 psi for commercial usage (National Ready Mixed Concrete Association, 2021)

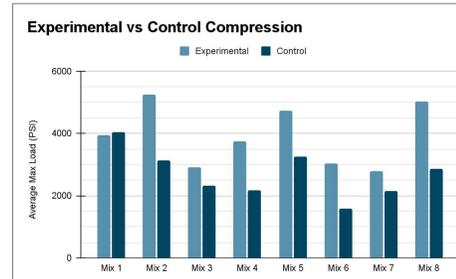


Figure 5. Experimental vs Control Compression Testing
Each bar shows the max load (PSI) sustained by each sample in a compression test until failure

Carbon Sequestration Results

- 57 samples tested in the first two trials (32 experimental, 25 control)

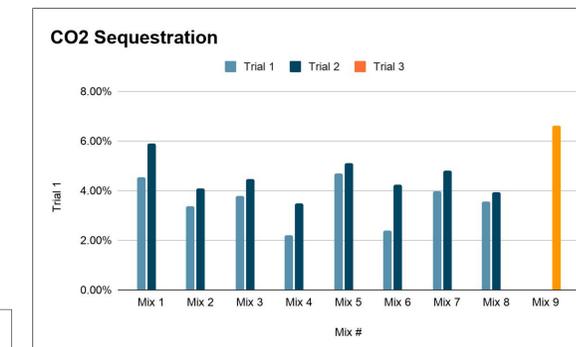


Figure 6. Carbon Sequestration by Mix

Discussion

- Mix optimization is essential for carbonation cured concrete. The optimized mix outperformed the initial eight mixes, suggesting that carbonation-cured concrete is suited to the iterative engineering workflow.
- Unclear pattern between the mixes which sequestered the most carbon and the mixes that gained the most compressive strength relative to their control counterparts.
 - Contradicts the expectation that the most carbonized mixes would increase in strength the most (Kazemian & Shafei, 2023)

Statistical Analysis

Mix	Mean CO ₂ %	SD	95% CI (lower, upper)	CI width
1	5.215	0.869	(3.833, 6.597)	2.764
2	3.740	0.732	(2.575, 4.905)	2.330
3	4.123	0.473	(3.371, 4.875)	1.504
4	2.835	0.868	(1.454, 4.216)	2.762
5	4.910	0.376	(4.312, 5.508)	1.196
6	3.305	1.153	(1.470, 5.140)	3.670
7	4.390	0.566	(3.489, 5.291)	1.802
8	3.523	0.485	(2.751, 4.295)	1.544
Optimal	6.600	0.0258	(6.559, 6.641)	0.082

Table 5: 95% Confidence Interval (CI) for CO₂ sequestration of each mix

Mix Optimal has smallest SD and the narrowest 95% CI by ~14.5x, showing consistent uptake

	SS	df	MS	F	p-value
Between Mixes	42.37155	8	5.29644	11.09586	8.661e-07

Figure 6: ANOVA testing results for CO₂ sequestration of each mix

Small P-value of 8.661e-7 indicates highly significant differences in CO₂% uptake across mixes

Tukey Post-HOC HSD Results:

Mix Optimal CO₂ sequestration significantly higher than all other mixes except Mix 1 (α = 0.05).
Mix 1 CO₂ sequestration not significantly higher than any other mixes (α = 0.05)

Conclusion

- Recycled materials were incorporated as SCM's in the concrete matrix, including steel slag and recycled aggregate, supporting sustainable construction and a circular economy
- Rapid CO₂ sequestration was achieved by the optimized mix, uptaking 6.60% of its mass in CO₂ in a 7-day curing period
- Consistent CO₂ sequestration was demonstrated in the optimized mix. Additionally, each sample of each mix retained the sequestered CO₂ between the end of carbonation and the final measurement.
- Structural Strength met or exceeded industry standards for 3 of the 8 initial mixes
- Applications: scalable process viable for controlled-production settings. Especially applicable for modular construction, which is predicted to grow into a trillion-dollar industry by 2040 (Bradley et al., 2024)



Future Work

- Quantify carbonation depth and improve depth of CO₂ penetration
- Evaluate long-term performance of carbonated concrete mix (freezing, rain)
- Qualify relationship between mix composition and strength difference between control and carbonated samples

Key References

Anabela, E. et al. "Real Time 3D Observations of Portland Cement Carbonation at CO₂ Storage Conditions." *Environmental Science & Technology*, vol. 54, no. 13, 11 June 2020, pp. 8323-8332. <https://doi.org/10.1021/acs.est.0c09798>. Accessed 10 Sept. 2023.

Bradley, C., Chui, M., Russell, K., Kwellin, E., Birshan, M., & Subay, C. (2024, October 23). The next big arena of competition. McKinsey & Company. <https://www.mckinsey.com/insight/research/the-next-big-arena-of-competition>

Elishabh Van Roijen, et al. "The Climate Benefits from Cement Carbonation Are Being Overestimated." *Nature Communications*, vol. 15, no. 1, 6 June 2024. <https://doi.org/10.1038/s41467-024-48965-z>. Accessed 16 Aug. 2024.

Fu, X., Guerni, A., Zampino, D., & Rotta Loria, A. F. (2024). Storing CO₂ while strengthening concrete by carbonating its cement in suspension. *Communications Materials*, 5(1), 1-14. <https://doi.org/10.1038/s43246-024-00544-9>

Kazemian, M., & Shafei, S. (2023). Carbon sequestration and storage in concrete: A state-of-the-art review of compositions, methods, and developments. *Journal of CO₂ Utilization*, 70, 102443. <https://doi.org/10.1016/j.cou.2023.102443>

Laurent, A. P. (2023, August 17). Here's how concrete can serve as a natural "carbon sink." *World Economic Forum*. <https://www.weforum.org/stories/2023/08/concrete-natural-carbon-sink-expert-explains/>

Li, N., Mo, L., & Jiljaner, C. (2022). Emerging CO₂ utilization technologies for construction materials: A review. *Journal of CO₂ Utilization*, 65, 102237. <https://doi.org/10.1016/j.cou.2022.102237>

National Ready Mixed Concrete Association. (2021). CIP 35: Testing compressive strength of concrete (Publication No. 35). <https://www.nrma.org/wp-content/uploads/2021/01/35.pdf>

Possan, E. et al. "CO₂ Uptake Potential due to Concrete Carbonation: A Case Study." *Case Studies in Construction Materials*, vol. 6, June 2017, pp. 147-161. <https://doi.org/10.1016/j.cscm.2017.03.007>. Accessed 20 Mar. 2020.

Pu, Y., et al. "Accelerated Carbonation Treatment of Recycled Concrete Aggregates Using Flue Gas: A Comparative Study towards Performance Improvement." *Journal of CO₂ Utilization*, vol. 43, Jan. 2021, p. 101362. <https://doi.org/10.1016/j.cou.2020.101362>. Accessed 9 Oct. 2021.

Ragipani, R., et al. "Direct Air Capture and Sequestration of CO₂ by Accelerated Indirect Airborne Mineral Carbonation under Ambient Conditions." *ACS Sustainable Chemistry & Engineering*, vol. 10, no. 24, 9 June 2022, pp. 7852-7861. <https://doi.org/10.1021/acscchem.1c07867>

Rodgers, L. (2018, December 17). Climate change: The massive CO₂ emitter you may not know about. BBC News. <https://www.bbc.com/news/science-environment-46455844>

Saturated Salt Solutions and Air Humidity. (2014). www.engineeringtoolbox.com/salt-humidity-d_1887.html

Shao, Y., et al. "Accelerated Carbonation of Portland Limestone Cement." *Journal of Materials in Civil Engineering*, vol. 26, no. 1, Jan. 2014, pp. 117-124. [https://doi.org/10.1061/\(ASCE\)1093-5533\(2007\)73](https://doi.org/10.1061/(ASCE)1093-5533(2007)73). Accessed 28 Oct. 2022.